

Turbulent combustion (Lecture 1) US	CViterbi School of Engineering
 Motivation Basics of turbulence Premixed-gas flames Turbulent burning velocity Regimes of turbulent combustion Flamelet models Non-flamelet models Flame quenching via turbulence Case study I: "Liquid flames" (turbulence without thermal expansion) Case study II: Flames in Hele-Shaw cells (thermal expansion without turbulence) 	
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Energy dissipation USC Viterbi
> Why is energy dissipated at small scales? > Viscous term in NS equation (v∇ ² u) at scale x ~ vu'/(x) ² > Convective term (u·∇)u ~ (u') ² /x > Ratio convective / viscous ~ u'x/v = Re _x > Viscosity more important as Re _x (thus x) decreases > Define energy dissipation rate (ε) as energy dissipated per unit mass (u') ² per unit time (τ ₁) by the small scales > ε ~ (u') ² /τ ₁ ~ (u') ² /(L ₁ /u') = C(u') ³ /L ₁ (units Watts/kg = m ² /s ³) > With suitable assumptions C ≈ 3.1 > Note to obtain u'/S _L = 10 for stoichiometric hydrocarbon-air mixture (S _L = 40 cm/s) with typical L ₁ = 5 cm, ε ≈ 4000 W/kg > Is this a lot of power or energy? > Turbulent flame speed S _T ≈ u' > Across distance L ₁ , heat release per unit time Q = Y _f Q _R S _T /L ₁ ≈ (0.068)(4.5 x 10 ⁷ J/kg)(4 m/s)/(0.05 m) ≈ 2.4 x 10 ⁸ W/kg > Dissipation rate / heat generation rate ε/Q ≈ 1.6 x 10 ⁻⁵ > Turbulent kinetic energy / thermal energy = (3/2 u' ²)/Y _f Q _R ≈ 8 x 10 ⁻⁶ ! > Answer: NO, very little power or energy
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Kolmogorov universality hypothesis (1941)

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USCViterbi Gaseous vs. liquid flames School of Engineering > Most model employ assumptions not satisfied by real flames Adiabatic (gas flames: sometimes ok) (Liquid flames TRUE!) \blacktriangleright Homogeneous, isotropic turbulence over many L_I (gas flames: never ok) (Liquid flames: can use different apparatuses where this is more nearly true) Low Ka or high Da (thin fronts) (gas flames: sometimes ok) (Liquid flames: more often true due to higher Sc) Lewis number = 1 (gas flames: sometimes ok, e.g. CH₄-air) (Liquid flames: irrelevant since heat transport not a factor in propagation) > Constant transport properties (gas flames: never ok, $\approx 25x$ increase in v and α across front!) (Liquid flames: TRUE) > u' doesn't change across front (gas flames: never ok, thermal expansion across flame generates turbulence) (but viscosity increases across front, decreases turbulence, sometimes almost cancels out) (Liquid flames: TRUE) Constant density (gas flames: never ok!) (Liquid flames: true, although buoyancy effects still exist due to small density change) Conclusion: liquid flames better for testing models! AME 513b - Spring 2020 - Lecture 6 - Turbulent premixed flames



Property	Stoichiometric hydrocarbon-air flame	Autocatalytic chemical front
Reaction mechanism	Many-step, chain-branching	Two-step, straight-chain
SL	40 cm/sec	0.03 cm/sec
$\beta = E/RT_{ad}$	10	0.05
Δρ/ρ _f	6	0.0003
$\Delta v / v_{R}$	25	0.02
Sc	1	500
Impact of heat loss	Critical	Irrelevant
Ease of LIF imaging	Tough (\$\$\$)	Trivial



























































USCViterbi **Propagation rates - orientation effect** School of Engineering \succ Upward - S_T/S_L \Downarrow as Pe \Uparrow (S_L increases, decreasing benefit of buoyancy); highest propagation rates > S_T/S_L converges to ≈ 3 at large Pe – same as horizontal 8 Upward; CH4-air: 0.5", 0.25", 0.125" 7 ◇ 0.5"* 6 0.5 × 0.25" 5 0.125" S_T/SL 4 $\overset{\diamond}{\exists}_{\diamond}$ 3 ₽¢ ⊟≝₀ 2 1 0 0 50 100 150 200 250 Pe AME 513b - Spring 2020 - Lecture 6 - Turbulent premixed flames













 Remark
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 > Most experiments are conducted in open flames (Bunsen, counterflow, ...) - gas expansion relaxed in 3rd dimension
 > ... but most practical applications in confined geometries, where unavoidable thermal expansion (DL) & viscous fingering (ST) instabilities cause propagation rates ≈ 3 S_L even when heat loss, Lewis number & buoyancy effects are negligible

 > DL & ST effects may affect propagation rates substantially even when strong turbulence is present - generates wrinkling up to scale of apparatus

 > (ST/S_L)Total = (ST/S_L)Turbulence X (ST/S_L)ThermalExpansion ?

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